## LITERATURE CITED

- 1. P.H. Scheuer (ed.), "Marine Natural Products," Vol. I, New York: Academic Press, 1978, and subsequent volumes in the series.
- 2. G.R. Schulte and P.J. Scheuer, Tetrahedron, 38, 1857 (1982).
- 3. A.F. Rose, P.J. Scheurer, J.P. Springer, and J. Clardy, J. Am. Chem. Soc., 100, 7665 (1978).
- 4. G.R. Pettit, Y. Kamano, P. Brown, D. Gust, M. Inoue, and C.L. Herald, J. Am. Chem. Soc., 104, 905 (1982).
- 5. A.A. Tymiak and K.L. Rinehart, Jr., J. Am. Chem. Soc., 105, 7396 (1983).
- 6. A. Grumbach, H. Schmid, and W. Bencze, Experientia, 8, 224 (1952).
- 7. H. Schmid and W. Bencze, Helv. Chim. Acta, 36, 205, 1468 (1953).
- 8. G. Alberts-Schönberg, W.V. Philipsborn, L.M. Jackman, and H. Schmid, *Helv. Chim. Acta*, 45, 1406 (1962).
- 9. A.J. Weinheimer, C.W.J. Chang, J.A. Matson, and P.N. Kaul, J. Nat. Prod., 41, 488 (1978).
- K. Nakanishi and P.H. Solomon, "Infarred Absorption Spectroscopy," San Francisco: Holden-Day, Inc., 1977, p. 47.

Received 13 July 1984

## FLAVONOIDS OF PARIETARIA OFFICINALIS

#### J. BUDZIANOWSKI,\* L. SKRZYPCZAK, and D. WALKOWIAK

Department of Pharmaceutical Botany, Karol Marcinkowski Medical Academy, 61-712 Poznań, Poland

The genus *Parietaria* (Urticaceae) comprises six species of which *Parietaria officinalis* L. is a medicinal plant (1,2). Previous chemical investigations indicated the presence of kaempferol-3-bioside and other unidentified flavonoids (3). This work is the first detailed report of flavonoids in the genus *Parietaria*.

The leaves and flowers of P. officinalis yielded ten flavonoids: the 3-glucosides and 3-rutinosides of quercetin, kaempferol and isorhamnetin, 3-sophorosides of quercetin and kaempferol, and 3-neohesperidosides of kaempferol and isorhamnetin.

#### **EXPERIMENTAL**

PLANT MATERIAL.—Flowering plants were collected in Poznań district, Poland. Voucher specimens are deposited at our department.

EXTRACTION AND ISOLATION.—Air-dried leaves and flowers (1000 g) were extracted five times with MeOH. Extracts were concentrated under reduced pressure, and the residue (145 g) was dispersed in hot  $H_2O$ . The  $H_2O$ -soluble portion was extracted with *n*-hexane, CHCl<sub>3</sub>, Et<sub>2</sub>O, EtOAc, and *n*-BuOH, successively. The two latter flavonoid-containing fractions (2.3 g and 8.0 g, respectively) were combined and chromatographed over a polyamide SC-6 column (100 g, 5×28.5 cm) with  $H_2O$  followed by an EtOH- $H_2O$  gradient. Three fractions obtained were further separated on preparative polyamide 6D plates (20×20 cm) with CHCl<sub>3</sub>-MeOH-methyl ethyl ketone, 9:4:2 (run twice). Bands were eluted with MeOH, and all individual compounds were finally purified over small Sephadex LH-20 columns in pure MeOH. Yields were low: 64 mg for quercetin-3-rutinoside, 20 mg for quercetin-3-sophoroside, 16 mg for isorhamnetin-3-neohesperidoside, and below 5 mg for the remaining compounds.

IDENTIFICATION OF THE FLAVONOIDS. —Flavonoids were identified by partial and total acidic hydrolysis, enzymatic hydrolysis,  $H_2O_2$  oxidation, uv and <sup>1</sup>H-nmr spectroscopy (4,5), analysis of methylated sugars obtained from permethylated compounds, and tlc comparisons with authentic samples when available. Uv spectra were taken on a uv-Vis Specord (Zeiss, Jena) and <sup>1</sup>H-nmr spectra were recorded on a JEOL FX 90 Q.

#### ACKNOWLEDGMENTS

We thank Prof. H. Itokawa, Tokyo College of Pharmacy, for the sample of isorhamnetin-3rutinoside and Doc. dr hab. K. Latowski, Adam Mickiewicz University, Poznań, for calling our attention to *P. officinalis*.

# **Brief Reports**

### LITERATURE CITED

- 1. G. Madaus, "Lehrbuch der biologischen Heilmittel, Heilpflanzen" Band 3, Leipzig: G. Thiem, 1938, p. 2062.
- 2. P. Fournier, "Encyclopedie Biologique 32, Plantes medicinales et veneneuses," ed. P. Lechevalier. Paris, 1948, p. 166.
- 3. J. Sotiropoulos, Arch. Pharm. (Athens), 24, 125 (1968); Chem. Abstr., 71, 84498 (1968).
- 4. J.B. Harborne, "Phytochemical Methods." London: Chapman and Hall, 1973, p. 73.
- 5. T.J. Mabry, K.R. Markham, and M.B. Thomas, "The Systematic Identification of Flavonoids," New York: Springer, 1970, p. 23.

Received 10 August 1984

### THE CHEMICAL CONSTITUENTS OF MUSANGA CECROPIOIDES

## Chukwunonye M. Ojinnaka<sup>1\*</sup>

Department of Chemistry, University of Port Harcourt, Nigeria

#### and JOSEPH I. OKOGUN

### Department of Chemistry, University of Ibadan, Nigeria

We wish to report the isolation, for the first time, of tormentic acid, 2-acetyl tormentic acid, 3-acetyl tormentic acid, and euscaphic acid, as their methyl esters, from the rootwood of *Musanga cecropioides* R. Br. (Urticaceae). These triterpene acids were previously isolated from *Myrianthus arboreus* P. Beauv (1,2). The two genera, *Musanga* and *Myrianthus*, which were formerly classified as members of the Moraceae (3,4), are now placed in the family Urticaceae because both genera lack latex (5). The presence of the same triterpene acids in both genera may be taxonomically significant.

### EXPERIMENTAL

PLANT MATERIAL.—*M. cecropioides* was collected from the bank of the New Calabar River near the University of Port Harcourt, Choba, Nigeria. Voucher specimens are deposited at the Herbarium of the Faculty of Science, University of Port Harcourt, Nigeria.

EXTRACTION AND ISOLATION.—The rootwood (15 kg) of M. cecropioides was crushed and successively extracted with hexane, CHCl<sub>3</sub>, and EtOH. The EtOH extract (58 g) was redissolved in 10% aqueous MeOH and precipitated with dilute HCl. The precipitate (42 g) was methylated (CH<sub>2</sub>N<sub>2</sub>), and the crude methyl ester (43 g) was chromatographed on a silica gel (1.2 kg) column eluted with Et<sub>2</sub>O in petroleum ether. Mixtures of 40%, 50%, 60%, and 70% Et<sub>2</sub>O in petroleum ether eluted 2-acetyl methyl tormentate, 3-acetyl methyl tormentate, methyl euscaphate, and methyl tormentate, respectively. These compounds were identified from their mp, ir, ms, and nmr and by direct chromatographic comparison with authentic samples isolated from Myriantbus arboreus (1,2).

Full details of the isolation, mps, spectral data, and identification of the compounds are available on request to the authors.

# ACKNOWLEDGMENTS

This research was supported by the University of Ibadan and the University of Port Harcourt Senate Research Grants. We are grateful to Mr. J.O. Opayemi for the collection of plant material and Mr. Ezumah Igwiloh for typing the manuscript.

## LITERATURE CITED

- 1. C.M. Ojinnaka, J.I. Okogun, and D.A. Okorie, Phytochemistry, 19, 2482 (1980).
- 2. C.M. Ojinnaka, J.I. Okogun, and D.A. Okorie, Phytochemistry, 23, 1125 (1984).
- 3. R.W.J. Keay, C.F.A. Onochie, and D.P. Stanfield, "Nigeria Trees," vol. 11, Ibadan, Nigeria: Department of Forest Research, 1964, p. 164.
- J.M. Dalziel, "The Useful Plants of West Tropical Africa," Millbank, London: Crown Agents for Overseas Government and Administrators, 1939, pp. 284-285.
- 5. J.C. Willis, "A Dictionary of the Flowering Plants and Ferns," 7th ed., Cambridge: University Press, 1966, pp. 748, 750.

Received 15 August 1984

<sup>1</sup>Present address: Department of Organic Chemistry, Arrhenius Laboratory, University of Stockholm, 106 91 Stockholm, Sweden.